

RAMAKRISHNA MISSION VIDYAMANDIRA

(Residential Autonomous College under University of Calcutta)

B.A./B.SC. FIFTH SEMESTER EXAMINATION, DECEMBER 2013

THIRD YEAR

CHEMISTRY (Honours)

Date : 19/12/2013

Time : 11 am – 1 pm

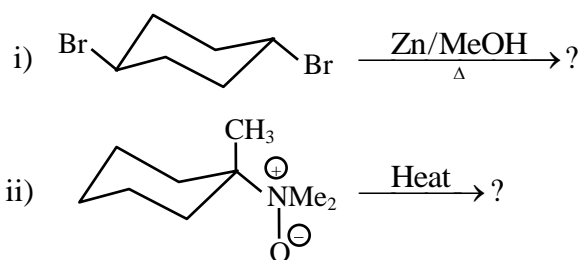
Paper : V-B

Full Marks : 50

[Answer one question from each unit]

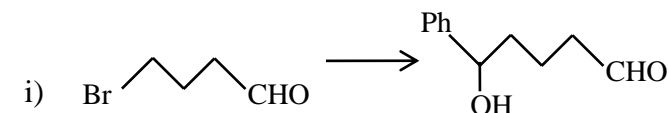
Unit - I

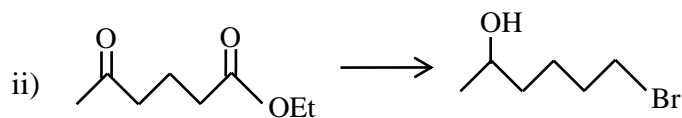
1. a) Draw the preferred conformation of 1-methyl-1-phenyl cyclohexane and justify your answer. [2]
b) Predict with reasons, which one of the following pair will undergo faster oxidation with chromic acid. [3]
trans-4-*t*-butylcyclohexanol and *cis*-4-*t*-butylcyclohexanol
c) Use Felkin-Anh model to explain the formation of major product : [3]
2. a) Draw the energy diagram for the ring inversion of cyclohexane following the C₂-pathway and explain the diagram. [3]
b) Explain that *trans*-4-*t*-butylcyclohexyl tosylate undergoes elimination to give 4-*t*-butylcyclohexene with the base ⁽⁻⁾SPh although not with much stronger base ⁽⁻⁾OEt . [3]
c) Predict the product(s) with mechanism : [3+3]



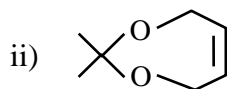
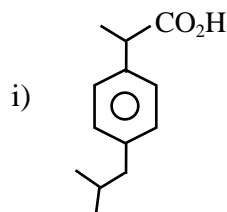
Unit - II

3. a) What is a synthon? Give one example of compounds that may form a² and d¹ types of synthons. [3]
b) Give retrosynthetic analysis and efficient synthesis for each of the following (any two) : [2½×2]
4. a) How would you carry out the following transformations? (any two) [2½×2]



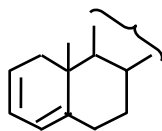


- b) Explain, with mechanism, the role of dicyclohexylcarbodiimide in the formation of the peptide linkage between two different suitably blocked amino acids. [3]
 c) Trace the route of synthesis of $\text{R-CH(NH}_2\text{)-CO}_2\text{H}$ from phthalimide. [2]
 d) Give retrosynthetic analyses of the following and show their forward synthesis also : (**any one**) [3]

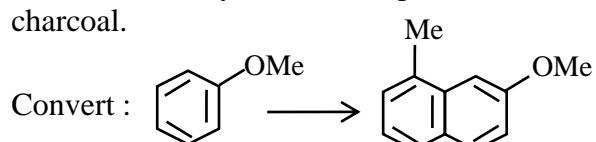


Unit - III

5. a) Define chemical shift and explain the terms—
 (i) downfield and (ii) upfield shifts as used in NMR spectroscopy. [3]
 b) How would you distinguish between the members of each of the following pairs of compounds as indicated. [2×3]
 i) *trans*-stilbene and *cis*-stilbene (by uv spectroscopy)
 ii) *ortho*-chloronitrobenzene and *para*-chloronitrobenzene (by NMR spectroscopy)
 iii) *ortho*-hydroxyacetophenone and *para*-hydroxyacetophenone (by IR spectroscopy)
 c) Draw the ^1H NMR signals of $\text{CH}_3\text{CH}_2\text{OH}$ (ordinary grade) showing relative chemical shifts, integration and spin-spin coupling patterns. [3]
 6. a) An organic compound [B] having molecular formula $\text{C}_4\text{H}_8\text{O}$ absorbs at 274 nm ($\epsilon_{\text{max}} = 17$) in uv spectroscopy. In IR, a strong absorption band is found at 1715 cm^{-1} , and medium absorption band are formed at $2940 - 2860\text{ cm}^{-1}(\text{m})$ and $1460\text{ cm}^{-1}(\text{m})$. The signals in NMR spectrum are (i) $\delta 2.48$ (quartet), (ii) $\delta 2.12$ (s) and (iii) $\delta 1.07$ (t).
 Assign the structure of [B], explaining the spectroscopic observations. [4]
 b) In n-hexane, mesityl oxide $[\text{Me}_2\text{C} = \text{CHCOMe}]$ shows two absorption peaks at λ_{max} 230nm and 270nm. Identify the electronic transition for each of them. Explain how these λ_{max} values shift if water is used as solvent instead of n-hexane. [3]
 c) Why TMS is used in NMR spectroscopy? [2]
 d) Calculate λ_{max} in uv spectrum for the following compound using Woodward-Hofmann's rule. [3]

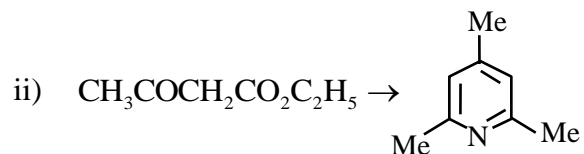
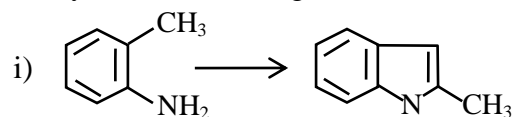


Unit - IV

7. a) Indole undergoes electrophilic substitution primarily at C-3 but pyrrole does at C-2. Explain. [2]
 b) How can you convert naphthalene to 2-aminonaphthalene? [3]
 c) Write the structure of the product showing the steps of reaction, when β -phenylethylamine is heated with acetyl chloride in presence of POCl_3 and the resulting product is heated with palladium-charcoal. [3]
 d) Convert :  [2]

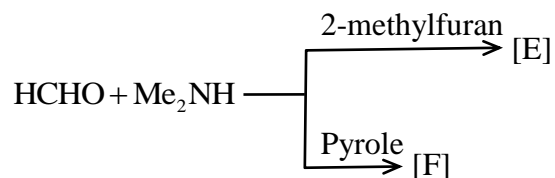
e) Outline the synthesis of ranitidine and mention one use of it. [3]

8. a) Carryout the following conversions. Write down the reaction conditions and the reagents involved. [2×2]



b) Convert : Phthalic acid \rightarrow 9-methylanthracene. [2]

c) Identify the products [E] and [F] in the following reactions and explain the reaction scheme : [4]



d) Outline the synthesis of nifedipine. Mention one use of it. [3]

